#### Atmospheric fluorescence

Fernando Arqueros Universidad Complutense de Madrid

## Outline

Introduction

- Features of air-fluorescence radiation
- The fluorescence technique for the detection of UHECR
- The atmospheric fluorescence
  - Molecular excitation by electron impact
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  - Dependence on atmospheric parameters
- The fluorescence yield
  - Definitions
  - The effect of secondary electrons
- Experimental techniques and results

#### 🔷 In Summary ..

# Introduction

## Cosmic Rays – Atmospheric Fluorescence What do they have to do?

- An ultra-high energy cosmic ray (>10<sup>18</sup>eV) generates a shower in the atmosphere containing a large number of charged particles, mainly electrons.
- Electrons lose energy by collision with atmospheric molecules. A very small fraction of these collisions excite/ionize nitrogen molecules to some specific levels which de-excite giving rise to UV light.
- Appropriate telescopes can detect this fluorescence light.

## HiRes, Auger, TA, ASHRA, JEM-EUSO, OWL

#### Features of the air-fluorescence radiation

- Near UV (300 400 nm).
- Emitted isotropically.
- Illustratively a  $10^{20}$  eV shower = a 100 W light bulb moving at the speed of light.

#### Detection of air showers using the fluorescence technique



# Fluorescence telescopes "see" the UV light emitted by $N_2$ molecules excited by shower electrons

## Measure of the EM energy of an air-shower using the fluorescence technique



Y(P,T,h) is measured in lab experiments

International effort to increase the accuracy of the fluorescence yield

#### 5<sup>th</sup> Fluorescence Workshop, El Escorial, Madrid – September 2007 http://top.gae.ucm.es/5th\_FW/

- Proceedings to be appear in Nucl. Instr. Methods A
- A summary of the workshop is already available at <u>arXiv:0807.3844</u>

## **Generation of air-fluorescence**

#### Molecular excitation by electron impact



#### **Radiative de-excitation**



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#### **Radiative de-excitation**

First Negative System (1N)  $N_2^+ (B^2 \Sigma_u^+ \rightarrow X^2 \Sigma_g^+)$ Second Positive System (2P)  $N_2 (C^3 \Pi_u \rightarrow B^3 \Pi_g)$ 





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Dependence on atmospheric parameters

#### Pressure dependence - Collisional de-excitation (quenching)

Excited nitrogen molecules may de-excite by collision with other molecules in the environment.

 $P = P' \implies$  collisional rate = radiative rate P' characteristic pressure Relative  $\frac{1}{P'} = \sum_{i} \frac{f_i}{P'_i}; \quad P_i' = \frac{kT}{\tau_r} \frac{1}{\sigma_{Ni} v_{Ni}} \overset{\text{relative}}{\swarrow}$  $I_{vv'}(P) \propto \frac{P}{1 + \frac{P}{P'}}$ Air components lifetime 🔨  $N_2$ ,  $O_2$ ,  $H_2O$ , Ar... Collisional cross section 337 nm <sup>-</sup>luorescence intensity 10 At high pressure P >> P' p'=151.7 hPa fluorescence intensity is 7.5 nearly P independent Fluorescence Intensity vs pressure provides a 2.5 measure of P' Pressure

Nagano et al. (2004)

#### **Apparent Lifetime**

#### AIRLIGHT (5th FW, EI Escorial 2007)



Lifetime of the population decreases with pressure P' can be measured from  $1/\tau$  versus P

#### **Temperature dependence**

$$\frac{1}{P'} = \sum_{i} \frac{f_{i}}{P'_{i}}; \quad P_{i}' = \frac{\mathbf{k}T}{\tau_{r}} \frac{1}{\sigma_{Ni} \overline{\nu}_{Ni}}$$
$$-\overline{\nu}_{Ni} = \sqrt{\frac{8kT}{\tau_{r}}}$$

Relative velocity grows with  $\sqrt{T}$ 

 $\pi\mu$ 

 $\sigma_{Ni} \propto T^{\alpha}$ 

Quenching cross section depends on the velocity of the colliders







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# The fluorescence yield

## **The fluorescence yield - Definitions**

 $\epsilon_{vv'}$  [m<sup>-1</sup>] Number of photons per unit electron path legth.

 $\frac{\varepsilon_{vv'}}{\rho} = \frac{A_{vv'}}{1 + P / P_v'} \qquad \text{Number of photons per unit column density (i.e. per gcm^{-2})} \\ A_{vv'} = \varepsilon_{vv'} / \rho \text{ in the absence of collisional quenching}$ 

#### Both $\varepsilon_{vv'}$ and $Y_{vv'}$ are measured in the laboratory

 $Y_{vv'}$ [Mev<sup>-1</sup>] Number of photons per unit deposited energy.

More useful for calorimetric applications

$$\frac{\mathrm{d} N_{ph}}{\mathrm{d} X} = Y(P, T, h) \frac{\mathrm{d} E_{dep}}{\mathrm{d} X} \qquad \qquad Y_{vv'} = \frac{Y_{vv'}^0}{1 + P / P_v'}$$

Relationship between  $Y_{vv'}$  and  $\varepsilon_{vv'}$ 

$$Y_{vv'} = \frac{\mathcal{E}_{vv}}{\left( \mathrm{d} E / \mathrm{d} x \right)_{\mathrm{dep}}}$$

## **Secondary electrons**



Fluorescence is mainly produced by secondary electrons ejected in ionization processes

Both fluorescence and deposited energy must be measured/computed in the same volume

Very important in lab experiments. Fluorescence from small volumes

## *E* [photons/m] vs.Y [photons/MeV]



#### Deposited energy is not equal to the energy loss for small volumes

# Experimental techniques and some results

## **Experimental techniques**



- Fluorescence measurement
  - Monochromators, filters
  - Photon counting: PMTs, HPDs Relationship between  $Y_{\mu\nu}$  and  $\varepsilon_{\mu\nu}$
- Gas properties: P, T, h, ...

Absolute calibration: number of <u>photons</u>, number of electrons, deposited energy

- $Y[MeV^{-1}]$  or  $\epsilon_{vv'}[m^{-1}]$
- Narrow (v-v') or wide spectral wavelength (e.g. 300 400 nm)

## Fluorescence yield versus electron energy

#### **Assumption:** Fluorescence yield is independent on electron energy,

Theoretical demonstration using a MC simulation which follow electrons down to a few eVs

#### AIRFLY (5th FW, EI Escorial 2007)

APS	6 – 30 keV
Argonne Wakefield Accelerator +	
Van de Graaff	0.5 – 15 MeV
BTF Frascati	50 – 420 Me\

Proportionality ( $\pm$  5%) <u>inside</u> E intervals. <u>Relative</u> calibration





#### MACFLY Astropart. Phys. (2007)



**FLASH (THICK TARGET) Astropart. Phys. (2006)** 5<sup>th</sup> FW, El Escorial 2007

## Absolute calibration with Rayleigh scattering



#### **Absolute calibration with Rayleigh scattering**



## **Absolute value of the Fluorescence Yield**

Experiment	E [MeV]	337 nm		Wide spectrum	
		m <sup>-1</sup>	MeV <sup>-1</sup>	m -1	MeV -1
AIRFLY (prel.)	350		4.12		
FLASH	2.85×10 <sup>4</sup>				20.8
MACFLY	1.5 - 5.0×10 <sup>4</sup>				17.6
Nagano et al.	0.85	1.02	5.03	3.81	
Lefeuvre et al.	0.85			4.23	
AIRLIGHT	0.25 – 2.0		5.68		
Kakimoto	1.4 - 10 <sup>3</sup>		5.7		

#### In summary

- Detection of air fluorescence provides a very useful tool for UHECRs detection.
  - Calorimetric measurement of the primary energy.
- ▶ The processes leading to the generation of fluorescence are well known.
  - The role of secondary electrons is very important
- Fundamental assumption: Fluorescence intensity proportional to deposited energy.
  - Theoretical and experimental tests.
- Accurate measurements of the dependence of the fluorescence yield on atmospheric parameters are being carried out. P'(T, h).
- Absolute measurements with uncertainties below 10% are being published.
  - Some disagreements.

#### More experimental results needed.

#### **Atmospheric Fluorescence**

Detailed information:

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